

## Biosynthetic Study of Betaenone B: Origin of the Oxygen Atoms and Accumulation of a Deoxygenated Intermediate using P-450 Inhibitor

Hideaki Oikawa,\* Akitami Ichihara, and Sadao Sakamura\*

Department of Agricultural Chemistry, Hokkaido University, Sapporo 060, Japan

The biosynthetic origin of oxygen atoms in betaenone B (**1**) was established by feeding experiment using [1-<sup>13</sup>C, <sup>18</sup>O<sub>2</sub>] acetate and by cytochrome P-450 inhibitor treatment of *Phoma betae*; incorporation patterns and accumulation of the plausible intermediate enabled us to propose a biosynthetic pathway for (**1**) involving an intramolecular Diels–Alder reaction at a late stage.

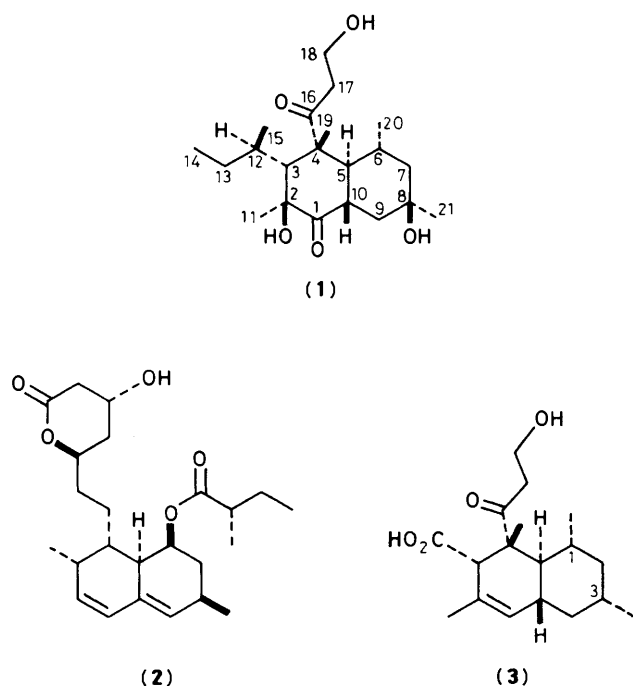
Hydroxylation is an important step at later stages of the biosynthesis of natural products, *i.e.* erythromycin,<sup>1</sup> rifamycin,<sup>2</sup> and trichothecene.<sup>3</sup> In this type of reaction the most common enzyme is cytochrome P-450, inhibitors for which have recently been reported.<sup>4</sup> These were used successfully to study the biosynthetic or metabolic pathways rather than their antifungal activity.<sup>5</sup> We now report the first application of cytochrome P-450 inhibitors for polyketide compounds.

Betaenone B (**1**) was isolated as the phytotoxin produced by *Phoma betae* Frank, and is a major metabolite in the betaenone family. In previous work, we showed that (**1**) was biosynthesized from eight acetate/malonate and five C<sub>1</sub>-units.<sup>7</sup> If we hypothesize that (**1**) is biosynthesized *via* an intramolecular Diels–Alder reaction, as for mevinolin (**2**) and other decalin polyketides,<sup>8</sup> the α-ketol moiety in (**1**) must be

derived from an alkene intermediate. In addition, the lack of a C-3 hydroxy group in diplodiatoxin (**3**)<sup>9</sup> implies that the C-8 hydroxylation in (**1**) occurs after ring formation. Therefore, we assume that (**1**) may be biosynthesized *via* an intermediate (**4**) (Scheme 1).

Incorporation of [1-<sup>13</sup>C, <sup>18</sup>O<sub>2</sub>]acetate was first carried out to examine the origin of the oxygen atoms. An isotopic shifted signal at C-16 (Δ+0.05 p.p.m.) was observed in the <sup>13</sup>C n.m.r. spectrum of the [1-<sup>13</sup>C, <sup>18</sup>O<sub>2</sub>]acetate-enriched (**1**), but not at C-1 and C-18. This result suggests that the oxygen atom at C-1 did not originate from acetate. The absence of an upfield shifted signal at C-18 is possibly due to washing-out of labelled oxygen at the carboxy group stage after hydrolysis of enzyme-bound polyketide.

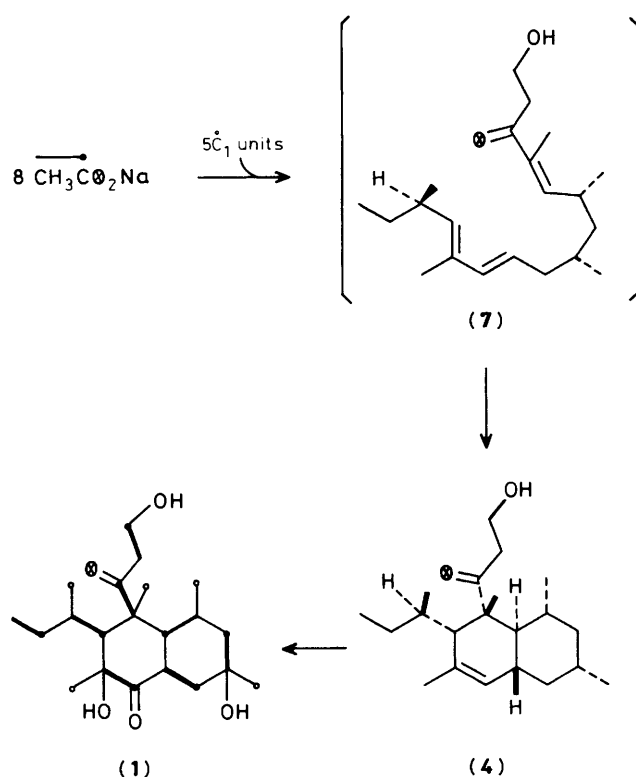
We then studied inhibition of biosynthesis. Thus, a solution



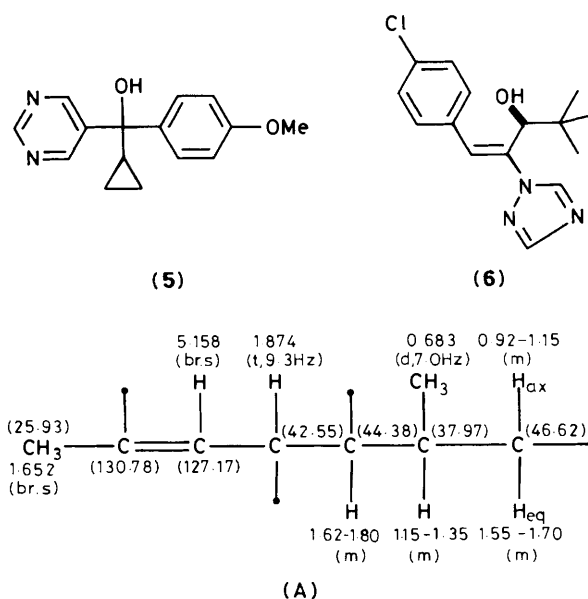
in dimethyl sulphoxide (DMSO) of ancymidol (**5**)<sup>10</sup> (5, 10, 20, and 40 mg, respectively) was added to four culture flasks containing 100 ml of 2% potato-sucrose medium after fermentation for 4 days. The production of (**1**) was suppressed in proportion to the concentration of (**5**) and a new metabolite (**4**) was detected in mycelial extracts. Treatment with (**5**) (40 mg) resulted in almost complete inhibition of formation of (**1**) to give 8.7 mg of (**4**) after SiO<sub>2</sub> chromatography. A similar result was obtained with the plant growth retardant, S-3307D (**6**).<sup>11</sup>

The intermediate (**4**), which we named probetaenone I, was isolated as a colourless oil: C<sub>21</sub>H<sub>36</sub>O<sub>2</sub>; [α]<sub>D</sub><sup>25</sup> -9.4° (c 2.5, MeOH). The <sup>13</sup>C n.m.r. spectrum of (**4**) showed 21 signals (CH<sub>3</sub>×6, CH<sub>2</sub>×5, CH×7, C×3). N.m.r. analysis (<sup>1</sup>H, <sup>13</sup>C, and <sup>1</sup>H-<sup>13</sup>C, <sup>1</sup>H-<sup>13</sup>C-COSY in C<sub>6</sub>D<sub>6</sub>) showed the presence of the partial structure (A), together with HO-CH<sub>2</sub>CH<sub>2</sub>C=O [δ<sub>C</sub> 58.04 (t), 41.40 (t), 215.03 (s); δ<sub>H</sub> 3.794 (ddd, *J* 4.4, 6.6, 11.4 Hz), 3.713 (ddd, *J* 4.4, 5.9, 11.4 Hz), 2.430 (ddd, *J* 4.4, 6.6, 18.3 Hz), 2.284 (ddd, *J* 4.4, 5.9, 18.3 Hz); ν<sub>max</sub> (neat) 3400 and 1695 cm<sup>-1</sup>], CH<sub>3</sub>-CH<sub>2</sub> [δ<sub>C</sub> 13.25 (q), 26.95 (t); δ<sub>H</sub> 0.890 (t, *J* 8.1 Hz)], CH<sub>3</sub>-C [δ<sub>C</sub> 17.78 (q), 53.85 (s); δ<sub>H</sub> 0.932 (s)], and 2 × CH<sub>3</sub>-CH [δ<sub>C</sub> 21.70 (q), 22.33 (q), 33.94 (d), 34.76 (d); δ<sub>H</sub> 0.855 (d, *J* 6.2 Hz), 0.918 (d, *J* 7.3 Hz)].<sup>†</sup> In addition to these data, the marked similarity of the <sup>1</sup>H n.m.r. spectrum of (**1**) to the spectra of (**2**) and (**3**), and the appearance of unusually upfield shifted methyl group (δ<sub>H</sub> 0.683, C-20), which is characteristic of betaenones, confirmed the structure of (**4**) as that shown in Scheme 1.

In conclusion, the inhibitor experiment suggests that (**4**) is an intermediate in the biosynthesis of (**1**) and the incorporation results are consistent with this. On the basis of these results, we propose a biosynthetic pathway for (**1**) (Scheme 1) in which an intramolecular Diels-Alder reaction of the triene (**7**) is probably a key step. Furthermore, cytochrome P-450 inhibitor might be useful for studies of biosynthetic pathways since it inhibits the oxidation, giving a less oxidized carbon skeleton. The synthesis of radioactive (**7**) is in progress.



Scheme 1



We thank Dr. K. Kamoshita, Sumitomo Chemical Co., Ltd., for a generous gift of S-3307D.

Received, 15th October 1987; Com. 1521

### References

- J. R. Martin and W. Rosenbrock, *Biochemistry*, 1967, **6**, 435.
- O. Ghisalba, P. Traxler, and J. Nüesch, *J. Antibiot.*, 1978, **31**, 1124.
- R. Evans, J. R. Hanson, and T. Marten, *J. Chem. Soc., Perkin Trans. 1*, 1976, 1212.

(References continued on p. 602)

<sup>†</sup> The <sup>13</sup>C n.m.r. signals of (**4**) were assigned from COSY (<sup>1</sup>H-<sup>1</sup>H and <sup>1</sup>H-<sup>13</sup>C) and chemical shift data, and by comparison with data for (**1**).

- 4 C. J. Coulson, D. J. King, and A. Wiseman, *Trends Biochem. Sci.*, 1984, **10**, 446.
  - 5 H. Wendorff and U. Matern, *Eur. J. Biochem.*, 1986, **161**, 391; F. VanMiddlesworth, A. E. Desjardins, S. L. Taylor, and R. D. Plattner, *J. Chem. Soc., Chem. Commun.*, 1986, 1156.
  - 6 A. Ichihara, H. Oikawa, K. Hayashi, S. Sakamura, A. Furusaki, and T. Matsumoto, *J. Am. Chem. Soc.*, 1983, **105**, 2907; A. Ichihara, H. Oikawa, M. Hashimoto, S. Sakamura, T. Haraguchi, and H. Nagano, *Agric. Biol. Chem.*, 1983, **47**, 2965.
  - 7 H. Oikawa, A. Ichihara, and S. Sakamura, *J. Chem. Soc., Chem. Commun.*, 1984, 814.
  - 8 R. N. Moore, G. Bigam, J. K. Chan, A. M. Hogg, T. T. Nakashima, and J. C. Vederas, *J. Am. Chem. Soc.*, 1985, **107**, 3694.
  - 9 P. S. Steyn, P. L. Wessels, C. W. Holzapfel, D. J. J. Potgieter, and W. K. A. Louw, *Tetrahedron*, 1972, **28**, 4775; A. Ichihara, H. Kawagishi, N. Tokugawa, and S. Sakamura, *Tetrahedron Lett.*, 1986, **27**, 1347.
  - 10 R. C. Coolbaugh, S. S. Hirano, and C. A. West, *Plant Physiol.*, 1978, **62**, 571.
  - 11 K. Izumi, Y. Kamiya, A. Sakurai, H. Oshio, and N. Takahashi, *Plant Cell Physiol.*, 1985, **26**, 821.
-